

Family list

15 family members for:

FR2154449

Derived from 14 applications.

- 1 1- TRIFLUOROMETHYL-PHENYL- 4-CYCLIC-AMIDOALKYL-PIPERAZINES USEFUL AS AMORECTIC AGENTS
Publication info: AR200387 A1 - 1974-11-08
- 2 1- TRIFLUOROMETHYL-PHENYL- 4-CYCLIC-AMIDOALKYL-PIPERAZINES USEFUL AS AMORECTIC AGENTS
Publication info: AT320655B B - 1975-02-25
- 3 1- TRIFLUOROMETHYL-PHENYL- 4-CYCLIC-AMIDOALKYL-PIPERAZINES USEFUL AS AMORECTIC AGENTS
Publication info: AT321307B B - 1975-03-25
- 4 TRIFLUOROMETHYLPHENYL PIPERAZINE DERIVATIVES
Publication info: AU4594072 A - 1974-02-28
- 5 1- TRIFLUOROMETHYL-PHENYL- 4-CYCLIC-AMIDOALKYL-PIPERAZINES USEFUL AS AMORECTIC AGENTS
Publication info: BE788280 A1 - 1973-02-28
- 6 1- TRIFLUOROMETHYL-PHENYL- 4-CYCLIC-AMIDOALKYL-PIPERAZINES USEFUL AS AMORECTIC AGENTS
Publication info: CH551430 A - 1974-07-15
- 7 1- TRIFLUOROMETHYL-PHENYL- 4-CYCLIC-AMIDOALKYL-PIPERAZINES USEFUL AS AMORECTIC AGENTS
Publication info: CH554899 A - 1974-10-15
- 8 1- TRIFLUOROMETHYL-PHENYL- 4-CYCLIC-AMIDOALKYL-PIPERAZINES USEFUL AS AMORECTIC AGENTS
Publication info: DE2242382 A1 - 1973-03-15
- 9 1- TRIFLUOROMETHYL-PHENYL- 4-CYCLIC-AMIDOALKYL-PIPERAZINES USEFUL AS AMORECTIC AGENTS
Publication info: ES406374 A1 - 1975-07-16
- 10 1- TRIFLUOROMETHYL-PHENYL- 4-CYCLIC-AMIDOALKYL-PIPERAZINES USEFUL AS AMORECTIC AGENTS
Publication info: FR2154449 A1 - 1973-05-11
- 11 1- TRIFLUOROMETHYL-PHENYL- 4-CYCLIC-AMIDOALKYL-PIPERAZINES USEFUL AS AMORECTIC AGENTS
Publication info: GB1368256 A - 1974-09-25
- 12 1-(TRIFLUOROMETHYL-PHENYL)-4-(CYCLIC-AMIDO-ALKYL) PIPERAZINES USEFUL AS ANORECTIC AGENTS
Publication info: IE37697 B1 - 1977-09-28
IE37697L L - 1973-03-04
- 13 1- TRIFLUOROMETHYL-PHENYL- 4-CYCLIC-AMIDOALKYL-PIPERAZINES USEFUL AS AMORECTIC AGENTS
Publication info: JP48085584 A - 1973-11-13
- 14 1- TRIFLUOROMETHYL-PHENYL- 4-CYCLIC-AMIDOALKYL-PIPERAZINES USEFUL AS AMORECTIC AGENTS
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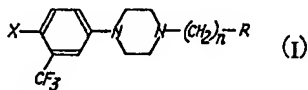
(72) Inventor PETER EDWARD CROSS

(54) 1-(TRIFLUOROMETHYL-PHENYL)-4-(CYCLIC-AMIDO-
 ALKYL)PIPERAZINES USEFUL AS ANORECTIC AGENTS

(71) We, PFIZER LIMITED, a British Company, of Ramsgate Road, Sandwich, Kent, do hereby declare the invention, for which we pray that a patent may be granted to us, and the method by which it is to be performed, to be particularly described in and by the following statement:—

The invention relates to compounds having anorectic properties and is particularly concerned with a class of novel 1 - (3 - trifluoromethyl - phenyl) - 4 - (cyclic - amido - alkyl) - piperazines which show anorectic properties with good duration of action and little or no central nervous system or cardiovascular activity, and show little tendency towards the development of drug tolerance. The compounds are therefore particularly useful in combatting a tendency towards obesity by reducing appetite in human subjects.

The compounds of the present invention are compounds of the general formula:—



where

R represents a succinimido, glutarimido, 2,4-dioxo - 1(or 3) - imidazolidinyl or 2,4 - dioxo - 1(or 3) - hexahydropyrimidinyl group, the last two groups being optionally substituted on the imino nitrogen atom with a methyl or an ethyl group;
 X represents a hydrogen, fluorine, chlorine or bromine atom; and
 n is 2 or 3;

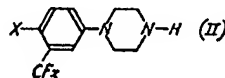
and the non-toxic acid addition salts or such compounds.

[Price 25p]

Non-toxic acid addition salts of the compounds of the invention can be prepared from acids which form addition salts containing non-toxic anions, such as the hydrochloride, hydrobromide, hydroiodide, sulphate or bisulphate, phosphate or acid phosphate, acetate, maleate, fumarate, oxalate, lactate, tartrate, citrate, gluconate, saccharate, and *p*-toluene sulphonate salts.

The compounds of the invention may be prepared in a number of ways:—

(1) A 1-aryl-piperazine of the formula:—



is reacted with an ω -R-substituted alkyl halide of the formula:



where hal represents a halogen atom, preferably a chlorine or a bromine atom, by heating in a dry inert organic solvent, e.g. dry dimethylformamide, in the presence of a base, e.g. potassium carbonate. Where the ω -R-substituted alkyl halide is the chloride or the bromide, the presence of an alkali metal iodide, e.g. potassium iodide, is advantageous.

The product may be isolated as the free base by addition of water to the cooled reaction mixture, extraction with a suitable organic solvent, e.g. diethyl ether, and evaporation of the previously washed organic solution *in vacuo* to dryness, or alternatively by evaporation of the reaction mixture *in vacuo* to dryness, extraction with a suitable organic solvent,



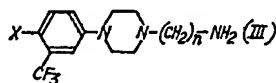
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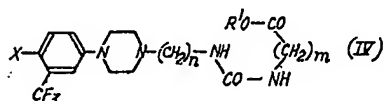
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e.g. diethyl ether, removal of undissolved residue from the solution by filtration, washing the filtrate with water and evaporation of the organic solution *in vacuo* to dryness. Purification of the crude product may then be effected in the usual manner by recrystallisation from a suitable solvent, e.g. petrol ether, to yield the free base, or by forming the acid addition salt, e.g. the hydrochloride, by addition of the appropriate acid in a suitable solvent, e.g. diethyl ether, to a solution of the crude base, e.g. in diethyl ether, and collection by filtration and recrystallisation of the precipitate from a suitable solvent, e.g. methanol, to produce the pure acid addition salt.

- (2) To prepare compounds of the invention in which R of the formula (I) represents the 2,4 - dioxo - 3 - imidazolidinyl or 2,4 - dioxo - 3 - hexahydropyrimidinyl group, an ω - (4 - aryl - 1 - piperazinyl) - alkylamine of the formula:—



is reacted with an alkyl isocyanatoacetate or β -isocyanatopropionate, the alkyl group preferably being a methyl or an ethyl group, by heating in a suitable reaction inert organic solvent, e.g. benzene, to produce a compound of the formula:—



where R¹ represents the alkyl group and m is 1 or 2. The product, of the formula (IV), may be isolated by evaporation of the reaction mixture *in vacuo* and purified by recrystallisation of the solid residue, formed if necessary from a gummy residue by trituration in e.g. petrol ether and decantation, from a suitable solvent, e.g. aqueous methanol solution.

Ring closure by elimination of the alcohol, R¹OH, to form the compound of the invention is then effected by heating the compound of the formula (IV), for example using as a heating source an oil bath at a temperature of about 200° C., for several hours, and the resultant solid mass is then cooled and purified by recrystallisation from a suitable solvent, e.g. a mixture of benzene and petrol ether, and optionally formed into the acid addition salt by conventional

means, as described in method (1), which is recrystallised, e.g. from methanol, to purity.

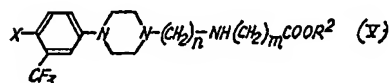
- (3) To prepare compounds of the invention in which R of the formula (I) represents a 2,4 - dioxo - 3 - methyl(or ethyl) - 1 - imidazolidinyl or 2,4 - dioxo - 3 - methyl(or ethyl) - 1 - hexahydropyrimidinyl group, an amine of the formula (III) is reacted with an alkyl haloacetate,



or β -halo-propionate,

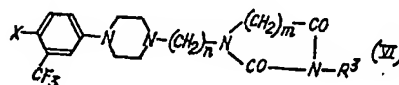


in which the alkyl group, R², and the halogen atom, hal, are preferably a methyl or an ethyl group and a chlorine or a bromine atom, respectively, in a suitable reaction inert organic solvent, e.g. benzene, in the presence of a base, e.g. triethylamine, at room temperature with stirring for several hours. The product, a compound of the formula:—



where R² is the alkyl group and m is 1 or 2, may be isolated by addition of water to the reaction mixture, basification e.g. with sodium hydroxide solution, separation of the organic layer, extraction of the aqueous solution with fresh solvent, e.g. benzene, combination of the organic solution retained and evaporation *in vacuo* to dryness. The resultant crude free base product may then be used directly in the final stage of the synthesis, or first purified by recrystallisation from a suitable solvent.

The free base product of the previous stage of the formula (V) is then reacted with methyl (or ethyl) isocyanate, R³NCO, in a suitable reaction inert organic solvent, e.g. dimethylformamide, at an elevated temperature to produce a compound of the formula:—



which may be isolated by evaporation *in vacuo* and purified as in method (1).

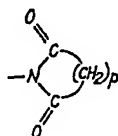
- (4) To prepare compounds of the invention in which R of the formula (I) represents

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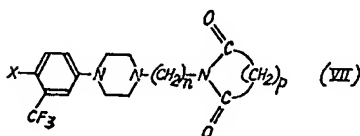
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a succinimido or glutarimido group, i.e. a group of the structure:—

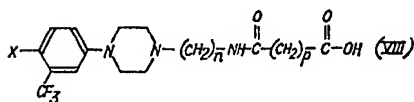


wherein p is 2 or 3, an amine of the formula (III) is reacted with succinic anhydride (p=2) or glutaric anhydride (p=3) either in solution in a suitable dry reaction inert organic solvent, e.g. dry pyridine, at an elevated temperature or in the absence of a solvent at an elevated temperature, e.g. at about 200° C. using an oil bath as a heating source, to effect the elimination of water between the reagents with production of the compound of the invention of the formula:—



The product may be isolated and purified by initial removal of solvent, if used, by evaporation *in vacuo* and in both cases recrystallisation of the reaction residue. If desired, an acid addition salt may then be prepared and purified in the conventional manner.

A by-product of the reaction, the corresponding succinic or glutaric half-amide of the formula:—



the presence of which can be detected by examination of the infrared absorption spectrum of the crude reaction product, may be formed, and in such a case it may be converted to the desired product of the formula (VII) by heating in boiling acetic anhydride, after which the solvent is removed by evaporation *in vacuo* and the residue purified as before.

The compound of the formula (III), used as a starting material in Methods (2) to (4), in which X represents a hydrogen atom and n is 2, is disclosed in First Addition No. 93884 to French Patent 1,537,901. That particular compound and others of the formula

(III) in which X and n are as hereinbefore defined may be prepared by the method described therein or analogous methods, by reacting unsubstituted or the appropriate halo-substituted 1 - (3 - trifluoromethylphenyl)piperazine with chloroacetonitrile and reducing the product.

Alternatively 1 - (3 - trifluoromethylphenyl)piperazine or its halo-substituted derivative may be reacted with either chloroacetamide or β-chloropropionamide, under reflux conditions in a reaction-inert organic solvent, e.g. ethanol and in the presence of a base, e.g. triethylamine, and the isolated product reduced to the desired 1 - aryl - 4 - (ω - aminoalkyl)piperazine of the formula (III) by use of, for example, sodium dihydro - bis[2 - methoxyethoxy] aluminate in benzene solution.

As a third alternative, the appropriate 1-aryl-piperazine may be reacted with ω-bromoethylamine or -propylamine as its hydrobromide salt under reflux conditions in a reaction-inert organic solvent, e.g. ethanol and in the presence of a base, e.g. sodium bicarbonate, and the product of the formula (III) isolated from the reaction mixture.

(5) Compounds of the invention in which X of the formula (I) represents a halogen substituent are prepared from compounds of the formula (I) in which X represents a hydrogen atom by appropriate nuclear aromatic halogenation using techniques well-known in the art.

(6) To prepare compounds of the invention in which X of the formula (I) represents a chlorine or a bromine atom, a compound of the formula (I) in which X represents a hydrogen atom is nitrosated such that the nitroso group is incorporated at the 4-position of the 3-trifluoromethylphenyl group, the nitroso group is reduced to an amino group, and the amino-substituted compound is submitted to diazotisation followed by the Sandmeyer reaction, using cuprous chloride or bromide, respectively, to yield the required chloro- or bromo-substituted final product. The nitrosation reaction is suitably performed by slowly adding an aqueous solution of sodium nitrite to a solution of the starting compound as a salt, e.g. the hydrochloride, in concentrated hydrochloric acid with vigorous stirring, the initial temperature of the mixture being kept low, e.g. at about -5° C. Generally a thick suspension will form, and the solid therefrom may be collected by filtration and used directly in the next stage.

Reduction of the 4 - nitroso - 3 - trifluoromethylphenyl compound to the corresponding 4 - amino - substituted compound is suitably effected by heating with tin and dilute hydrochloric acid under

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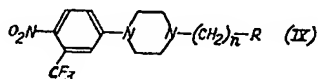
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reflux conditions for several hours, although several other reductive methods are possible, e.g. using iron or zinc with hydrochloric or acetic acid, zinc or titanium chloride with hydrochloric acid, and hydrogenation in the presence of such catalysts as Raney nickel, platinum or platinum oxide at room temperature and atmospheric or somewhat higher pressures. The product is isolated by removing sediment from the cooled reaction mixture, basifying the solution, e.g. by addition of aqueous sodium hydroxide solution, and extracting the product into an organic solvent, e.g. diethyl ether, and finally evaporating the solution *in vacuo* to dryness. Purification of the product may be effected by recrystallisation if desired, or the crude product may be used directly in the next stage. Diazotisation is conveniently performed by slowly adding an aqueous solution of sodium nitrite portionwise to a cooled solution of the amino compound in hydrochloric or sulphuric acid until it has been established, e.g. by a positive test with starch-iodide indicator paper, that excess nitrous acid is present. The resulting solution of the diazonium chloride or sulphate, respectively, may then be used as such in the subsequent Sandmeyer reaction.

For the preparation of the chloro compound by the Sandmeyer procedure, a solution of one equivalent of cuprous chloride in hydrochloric acid is added to the ice-cold diazotisation reaction solution, the diazotisation in this case having been performed using hydrochloric acid. The sparingly soluble complex which separates is decomposed by warming the mixture, nitrogen thereby being evolved, and thereafter the product is suitably isolated by dilution of the solution with water, extraction into a suitable organic solvent, e.g. diethyl ether, evaporation of the organic solution *in vacuo* to dryness and recrystallisation of the solid residue to purity. Optionally a desired acid addition salt may be prepared by conventional techniques. The bromo compound may be prepared by a similar procedure, but starting from a sulphuric acid diazotisation reaction solution and a solution of cuprous bromide in hydrobromic acid.

(7) Compounds of the invention in which X of the formula (I) represents a chlorine or a bromine atom may also be prepared from compounds of the formula:—

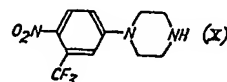
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by reduction of the nitro group to an amino group under conditions similar to those described for the reduction of the nitroso group to an amino group in Method (6), followed by diazotisation and the Sandmeyer reaction to effect the conversion of the amino group to a 4-chloro- or 4-bromo-substituent, as also described in Method (6). The starting compound of the formula (IX) is conveniently prepared by one of the three following methods:

(a) A compound of the formula (I) in which X represents a hydrogen atom is nitrated with a nitrating mixture comprising concentrated nitric and sulphuric acids, and the required 4-nitro - 3 - trifluoromethylphenyl compound is separated if necessary from other nitrated derivatives, e.g. by fractional crystallisation.

(b) A 1-aryl-piperazine of the formula:—

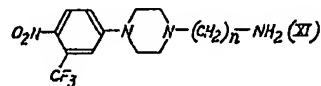


is reacted with an ω -R-substituted alkyl halide of the formula:



where hal represents the halogen atom, preferably a chlorine or a bromine atom according to the conditions as hereinbefore described in Method (1).

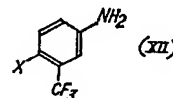
(c) An ω - (4 - aryl - 1 - piperazinyl)-alkylamine of the formula:—



(prepared by analogous methods to those hereinbefore described for the preparation of compounds of the formula (III))

is submitted to any of the preparative procedures of Methods (2), (3) and (4) as described for the corresponding des-nitro compound of the formula (III).

(8) All the compounds of the invention may be prepared by reacting an aniline derivative of the formula:—

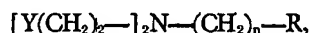


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with a compound of the formula:



in which Y represents a hydroxyl group or a halogen atom, preferably a chlorine atom, in a suitable reaction inert organic solvent, e.g. benzene, in the presence of an acid or base according to whether Y is a hydroxyl group or a halogen atom, respectively.

Alternatively, for the preparation of compounds of the invention of the formula (I) in which X represents a chlorine or a bromine atom, a starting material of the formula (XII) with X replaced by a nitro group may be utilised, and the product of the formula (IX) may be converted to the required final product by means of the procedure indicated in Method (7), as described in greater detail in Method (6).

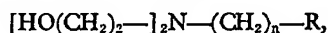
The starting material of the formula:



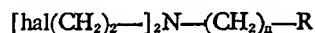
is conveniently prepared by reacting diethanolamine with an ω -R-substituted alkyl halide of the formula:



where hal represents a halogen atom, preferably a chlorine or a bromide atom, under conditions similar to those described in Method (1) for the reaction of such a compound with a 1-aryl-piperazine of the formula (II). The product, a compound of the formula:

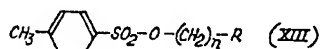


may optionally then be converted to the compound of the formula:



by conventional techniques for the conversion of a primary hydroxyl group to a halogen atom.

(9) As a variation on Method (1) hereinbefore described, the ω -R-substituted alkyl halide, $hal-(CH_2)_n-R$, may be replaced by the tosylate of ω -R-substituted alkanol, of the formula:—



and this may be reacted with a 1-aryl-piperazine of the formula (II) under similar conditions.

Preparation of the compound of formula II in which X is hydrogen, and compounds of formula III in which X is halogen, which are used as starting materials for methods (1) and (9), is described in British patent specification No. 948,767 (United States patent specification No. 3,170,926) and in United States patent specification No. 3,637,705, respectively.

The preparation of compounds of the present invention is described in the following Examples, in which all temperatures are given in °C.

EXAMPLE I.

1 - (3 - Trifluoromethylphenyl)piperazine (11.5 g) was added to a mixture of 2-succinimidoethyl chloride (8.1 g), anhydrous potassium carbonate (7.0 g) and potassium iodide (2.0 g) in dry dimethylformamide (50 ml). The mixture was then heated to 100° and maintained at that temperature for 24 hours, after which it was cooled and poured into water (250 ml). The aqueous solution was extracted with diethyl ether (3 × 100 ml) and the organic layers combined, washed with water and evaporated *in vacuo* to give a brown oil which subsequently crystallised. Recrystallisation from 80—100° petrol ether afforded a cleaner crystalline product from which the hydrochloride salt was prepared by addition of ethereal hydrogen chloride solution to an ethereal solution of the free base and collection by filtration of the resultant precipitate. The salt was recrystallised from a mixture of methanol and 2-butanone to yield 9.1 g of pure 1 - (2 - succinimidoethyl) - 4 - (3 - trifluoromethylphenyl) - piperazine hydrochloride as white crystals, m.p. 239—241°.

Analysis:—

Found:

C, 51.9; H, 5.5; N, 10.7%

Required for $C_{17}H_{20}F_3N_3O_2 \cdot HCl$:

C, 52.1; H, 5.4; N, 10.7%

This compound has been found to be embryotoxic and teratogenic in tests in pregnant rats and mice, but not in rabbits.

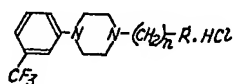
EXAMPLES II and III.

By methods similar to that of Example I, the compounds shown in the following table were prepared from 1 - (3 - trifluoromethylphenyl) - piperazine and the appropriate ω -cyclic - amido - alkyl chloride.

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Example	R	n	m.p. °C	Analysis % (Theoretical in brackets)		
				C	H	N
II	succinimido	3	233—5°	51.1 (51.1)	5.5 5.9	10.0 9.9
III	glutarimido	2	195—200°	52.7 (53.0)	5.9 5.7	9.7 10.3

EXAMPLE IV.

To a solution of 1 - (2 - aminoethyl) - 4 - (3 - trifluoromethyl phenyl) - piperazine (5.5 g) in benzene (50 ml) was added ethyl isocyanatoacetate (2.6 g) and the mixture was refluxed for one hour, allowed to stand at room temperature overnight and evaporated to dryness *in vacuo*. The residual gummy solid was triturated in 40—60° petrol ether, the solvent decanted and the solid recrystallised from aqueous methanol solution to afford 1 - [2 - (3 - {ethoxycarbonylmethyl}ureido)ethyl] - 4 - (3 - trifluoromethyl - phenyl) - piperazine (7.1 g) as white crystals, m.p. 126—8°.

Analysis:—

Found:

C, 54.0; H, 6.3; N, 14.0%
 Required for $C_{18}H_{22}F_3N_4O_3$:
 C, 53.7; H, 6.25; N, 13.9%

The product of the previous stage (5.3 g) was heated in an oil bath to a temperature of 180° and kept at that temperature for 2 hours. The resultant glassy solid was cooled and recrystallised from a mixture of benzene and petrol ether to give a cream-coloured solid. Formation of the hydrochloride salt of the product was effected in the usual manner and this was recrystallised from methanol to yield 3.3 g of 1 - [2 - (2,4 - dioxo - 3 - imidazolidinyl)ethyl] - 4 - (3 - trifluoromethylphenyl) - piperazine hydrochloride, m.p. 300—310°.

Analysis:—

Found:

C, 49.2; H, 5.1; N, 14.1%
 Required for $C_{18}H_{19}F_3N_4O_2 \cdot HCl$:
 C, 48.9; H, 5.1; N, 14.3%

EXAMPLE V.

To a stirred mixture of 1 - (2 - aminoethyl) - 4 - (3 - trifluoromethylphenyl) - piperazine (5.46 g) and triethylamine (0.04 g) in benzene (70 ml) was added ethyl bromoacetate (3.34 g) whereupon precipitation of white solid occurred. Stirring at room temperature was continued for 14½ hours, after which water was added, the aqueous layer basified by addition of aqueous sodium hydroxide solution, and the benzene layer separated from the aqueous layer. The latter was extracted with fresh benzene, and the combined benzene solutions evaporated *in vacuo* to yield a colourless oil (7.2 g).

A portion of the oil was converted to the hydrochloride salt of the product by conventional means which was recrystallised in turn from mixtures of methanol and isopropanol, methanol and 2-butanone and from ethanol to afford 1 - [2 - (ethoxycarbonylmethyl)aminoethyl] - 4 - (3 - trifluoromethylphenyl) - piperazine dihydrochloride, m.p. 199—201° as white crystals.

Analysis:—

Found:

C, 46.6; H, 6.4; N, 9.7%
 Required for $C_{17}H_{21}F_3N_3O_2 \cdot 2HCl$:
 C, 47.2; H, 6.1; N, 9.7%

A mixture of the crude free base product of the previous stage (5.39 g) and methyl isocyanate (1.2 g) in dimethylformamide (50 ml) was stirred at 60° for 6 hours. The mixture was then cooled and water added, whereupon an oil partially separated. Attempts to dissolve the oil in a reasonable quantity of chloroform failed and so the whole was evaporated *in vacuo* to an oil from which the hydrochloride salt was formed in the usual manner. The salt

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was recrystallised in turn from ethyl acetate containing a trace of methanol and from a mixture of *iso*-propanol and methanol to yield 1.5 g of 1 - [2 - (2,4 - dioxo - 3 - methyl-1 - imidazolidinyl)ethyl] - 4 - (3 - trifluoromethylphenyl) - piperazine hydrochloride as white needles, m.p. 231—2°.

Analysis:—

Found:

10 C, 50.37; H, 5.37; N, 13.62%
Required for $C_{17}H_{21}F_3N_5O_2 \cdot HCl$:
C, 50.31; H, 5.22; N, 13.8%

EXAMPLE VI.

15 1 - (4 - Chloro - 3 - trifluoromethylphenyl)-piperazine (5.0 g) was added to a mixture of 2-succinimidoethyl chloride (3.9 g), anhydrous potassium carbonate (2.65 g) and potassium iodide (0.75 g) in dry dimethylformamide (50 ml), and the mixture heated to 100° C. at which it was maintained for 19 hours. The solution was thereafter cooled, poured into water (250 ml) and the aqueous solution extracted with diethyl ether (3 × 100 ml), the organic layers then being combined, washed with water and evaporated *in vacuo* to afford an oil which subsequently crystallised. Re-

crystallisation from 80—100° petroleum ether afforded pale yellow crystals from which the hydrochloride salt was prepared by addition of ethereal hydrogen chloride solution to an ethereal solution of the free base and collection of the resultant precipitate by filtration. The salt was recrystallised from a mixture of methanol and 2-butanone to yield 1.8 g of pure 1 - (4 - chloro - 3 - trifluoromethylphenyl) - 4 - (2 - succinimidoethyl)piperazine hydrochloride, m.p. 256—7°.

Analysis:—

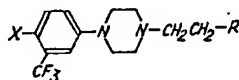
Found:

40 C, 48.1; H, 4.7; N, 9.7%
Required for $C_{17}H_{19}ClF_3N_5O_2 \cdot HCl$:
C, 47.9; H, 4.7; N, 9.8%

This compound has been found to be embryotoxic and teratogenic in tests in pregnant rats.

EXAMPLES VII AND VIII.

By methods similar to that described in Example I, the compounds shown in the following Table were prepared from 1 - (4 - halo-3 - trifluoromethylphenyl)piperazine and the appropriate ω - cyclic - amidoalkyl chloride.



Example	X	R	m.p. °C and free base/ salt form	Analysis % (Theoretical in brackets)		
				C	H	N
VII	Cl	2,4-dioxo-1-methyl-3-imidazolidinyl	227—8° (dihydrochloride)	42.5 (42.75)	4.6 4.64	12.0 11.7)
VIII	Br	succinimido	122—3° (free base)	47.02 (47.00)	4.39 4.37	10.14 9.67)

EXAMPLE IX.

55 To a stirred solution of 1 - (2 - aminoethyl) - 4 - (3 - trifluoromethylphenyl)piperazine (2.73 g) in 50% aqueous acetic acid (50 ml) at room temperature was added dropwise a solution of bromine (1.75 g) in 50% aqueous acetic acid (15 ml). Within 5 minutes from completion of bromine addition the solution, had turned colourless from its initial red colour. Stirring at room temperature was continued for a further ½ hour, and thereafter the solution was allowed to stand for one week.

65 The solution was then evaporated *in vacuo*

to remove solvent, and the crude product basified by addition of aqueous sodium hydroxide solution, the whole then being extracted with diethyl ether several times. The ethereal extracts were combined, dried over anhydrous sodium sulphate and evaporated *in vacuo* to an oil (about 3.5 g).

By a conventional technique, the hydrochloride salt of the product was prepared from the oil, and recrystallised from methanol to afford 2.0 g of 1 - (2 - aminoethyl) - 4 - (4-bromo - 3 - trifluoromethylphenyl)piperazine, m.p. 226—8° with decomposition.

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*Analysis:—**Found:*

C, 34.02; H, 4.52; N, 9.00%

Required for $C_{10}H_{17}BrF_3N_3 \cdot 3HCl$:

5 C, 33.83; H, 4.37; N, 9.10%

10 A mixture of the amine product of the previous stage (3.3 g, free base, isolated from the hydrochloride salt by basification) and succinic anhydride (0.94 g) was heated by means of an oil bath to 190° and then allowed to cool to room temperature. During the heating, evolution of gas was observed, and the oil reaction mixture darkened. The crude product was crystallised once from absolute ethanol to afford two crops (0.9 g and 0.4 g) of crystalline product, m.p. 123—123.5° for the first crop.

20 A mixed melting point of 123—124° was observed for the mixture of this product (first crop) with the product of Example VIII, and comparison of infrared spectra established its identity with the product of Example VIII.

25 The compounds of the invention have been found to be potent anorectic agents with advantages over those currently in use.

30 This has been shown in tests in which their anorectic effect has been measured in rats. In one of such tests the appetite for peeled potatoes of a group of rats starved for 18 hours before oral administration of 10 mg/kg of the test compound (as the free base) and then allowed access to the potatoes half an hour later, measured after periods of 2 hours and 5 hours from the time of presentation of their diet, was compared with the appetite of a second group of control rats which had been subjected to the same diet restrictions but to which only the vehicle for the test compound was administered, usually distilled water. The first group showed a considerably reduced intake of potato compared with the second (control) group after 2 hours, a situation not reversed during the final 3 hours. The 2 hour result demonstrated the presence of anorectic activity of the particular test compound administered whilst the 5 hour result demonstrated its duration of activity. In a second, similar test, various doses were administered in order to calculate one which caused the reduction of food intake by 50% compared with that of the controls at 2 hours (ED₅₀ 2 hours) and one producing this effect at 5 hours (ED₅₀ 5 hours). A comparison between the ED₅₀ values at 2 and 5 hours, as in the primary screen previously described, gave an indication of the duration of action of the test compound.

60 In order to show the absence or otherwise of central nervous system stimulation or sedation, rats were given the test compound orally 2 hours before being placed individually in compartments of an activity recorder and their locomotor activity was measured by counting electronically the interruptions of 2

65 narrow light beams passing between sources and photoelectric cells along the bottom of each compartment over a period of ten minutes. The average result for 12 animals was compared with that for control animals in the same time period. Dosages were varied in order to calculate that required to increase or decrease by 50% the locomotor activity compared with the value obtained using control animals (ED₅₀ 2 hours). The ED₅₀ (2 hours) values for anorexia and locomotor activity were then compared, a ratio

$$\frac{ED_{50} \text{ locomotor activity}}{ED_{50} \text{ anorexia}} \quad (\text{each at 2 hours})$$

greater than 12 being taken to indicate a selective anorectic effect unassociated with any central nervous system excitation or depressant effect of the test compound. Little drug tolerance was shown by the rats to the compounds administered at a rate of 10 mg/kg/day over a period of several weeks in that anorectic activity was maintained at a high level over the period after only a slight decrease during the first week of treatment.

90 The results of such testing have shown that compounds having the formula (I) where n is 2 and those having the formula (I) where R represents a succinimido group constitute preferred classes of compounds. Compounds of the invention having both such features in the formula (I) are particularly preferred. Amongst the very best compounds of the invention are 1 - (2 - succinimidoethyl) - 4 - (3 - trifluoromethylphenyl)piperazine and 1 - (2 - succinimidoethyl) - 4 - (4 - chloro - 3 - trifluoromethylphenyl)piperazine, i.e. the compounds of Examples I and VI, respectively.

100 The compounds of the invention can be administered alone, but will generally be administered in admixture with non-toxic carrier or diluent selected with regard to the intended route of administration and comparable with standard pharmaceutical practice. For example, they may be administered orally in the form of tablets containing such excipient as starch or lactose, or in capsules either alone or in admixture with excipients, or in the form of elixirs or suspensions containing flavouring or colouring agents. They may be injected parenterally, for example, intramuscularly or subcutaneously. For parenteral administration, they are best used in the form of a sterile aqueous solution which may contain other solutes, for example, enough salts or glucose to make the solution isotonic.

120 For administration to a human subject for the purpose of combatting a tendency towards obesity by reducing the appetite, it is expected that oral dosage of the compounds of the invention will be in the range from 0.01 to 10 mg/kg/day, more probably 0.1 to 1, given 125

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in a single dose or in divided doses. Thus, for typical adult patients, weighing from 50 to 80 kg, individual tablets or capsules, for administration once a day, or up to 4 times a day, could contain from 1 to 200 mg, more probably 5 to 80 mg, of active constituent in a suitable vehicle or carrier. The physician in any event will determine the actual dosage which will be most suitable for an individual patient and it will vary with age, the weight and response of that patient.

Suitable capsule dosage forms of compounds according to the invention are illustrated by the following Example:—

15 EXAMPLE X.

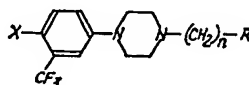
The product of Example I, 1 - (2 - succinimidoethyl) - 4 - (3 - trifluoromethylphenyl)-piperazine hydrochloride, was converted to the free base and formulation of capsules of this compound was then effected using the following constituents:—

Capsule (A)		mg/capsule
1 - (2 - succinimidoethyl) - 4 - (3 - trifluoromethylphenyl)-piperazine (free base, active constituent)	5	
Maize starch	110	
Lactose	225	
Lubricant (9 parts magnesium stearate to 1 part sodium lauryl sulphate)	8	
Total weight of constituents	348	
Capsule (B)		
1 - (2 - succinimidoethyl) - 4 - (3 - trifluoromethylphenyl)-piperazine (free base), active constituent	20	
Maize starch	110	
Lactose	250	
Lubricant (9:1 magnesium stearate:sodium lauryl sulphate)	8	
Total weight of constituents	388	
Capsule (C)		
1 - (2 - succinimidoethyl) - 4 - (3 - trifluoromethylphenyl)-piperazine (free base, active constituent)	40	
Maize starch	100	
Lactose	200	
Lubricant (9:1 magnesium stearate:sodium lauryl sulphate)	8	
Total weight of constituents	348	

In each case the active constituent was blended with the maize starch, lactose and half the lubricant (the latter having been screened through a 60 mesh screen), and the mixture was compressed to a granular state passable through a 30 mesh screen. The rest of the lubricant was then blended in and the mixture was filled into hard gelatin capsules of suitable size.

WHAT WE CLAIM IS:—

1. Compounds having the formula:



where

- R represents a succinimido, glutarimido, 2,4 - dioxo - 1 (or 3) - imidazolidinyl or 2,4 - dioxo - 1(or 3) - hexahydropyrimidinyl group, the last two groups being optionally substituted on the imino nitrogen atom with a methyl or an ethyl group; X represents a hydrogen, fluorine, chlorine or bromine atom; and n is 2 or 3; and their non-toxic acid addition salts.
- Compounds as claimed in Claim 1, in which X represents a hydrogen atom.
- Compounds as claimed in Claim 1, in which X represents a fluorine, chlorine or bromine atom.
- Compounds as claimed in any preceding claim, in which n is 2.
- Compounds as claimed in any preceding claim, in which R represents a succinimido group.
- 1 - (2 - succinimidoethyl) - 4 - (3 - trifluoromethylphenyl)piperazine and its non-toxic acid addition salts.
- 1 - (2 - succinimidoethyl) - 4 - (4-chloro - 3 - trifluoromethylphenyl)piperazine and its non-toxic addition salts.
- A compound as claimed in any preceding claim the preparation of which is described in any one of the Examples.
- A composition comprising a compound as claimed in any preceding claim in admixture with a non-toxic carrier or diluent.
- A tablet or capsule containing from 5 to 80 mg of a compound as claimed in any of claims 1 to 8 in admixture with a non-toxic carrier material.
- A method of reducing appetite in a human subject, and thereby combatting a tendency towards obesity, comprising administering to a human subject an effective amount of a compound as claimed in any of claims 1 to 8.
- A method as claimed in claim 11, comprising administering from .01 to 10 mg of

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the compound per kg weight of the subject per day. the compound per kg weight of the subject per day. 5

13. A method as claimed in claim 12, comprising administering from 0.1 to 1 mg of

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